

## Polyatomic Cations and Anions to Know

Cations					
ammonium	$\text{NH}_4^+$	hydronium	$\text{H}_3\text{O}^+$	mercury (I)	$\text{Hg}_2^{2+}$
Anions					
carbonate	$\text{CO}_3^{2-}$	hydrogen carbonate	$\text{HCO}_3^-$	perchlorate	$\text{ClO}_4^-$
chlorate	$\text{ClO}_3^-$	chlorite	$\text{ClO}_2^-$	hypochlorite	$\text{ClO}^-$
chromate	$\text{CrO}_4^{2-}$	dichromate	$\text{Cr}_2\text{O}_7^{2-}$	cyanide	$\text{CN}^-$
hydroxide	$\text{OH}^-$	iodate	$\text{IO}_3^-$	permanganate	$\text{MnO}_4^-$
nitrate	$\text{NO}_3^-$	nitrite	$\text{NO}_2^-$	peroxide	$\text{O}_2^{2-}$
phosphate	$\text{PO}_4^{3-}$	hydrogen phosphate	$\text{HPO}_4^{2-}$	dihydrogen phosphate	$\text{H}_2\text{PO}_4^-$
sulfate	$\text{SO}_4^{2-}$	hydrogen sulfate	$\text{HSO}_4^-$	sulfite	$\text{SO}_3^{2-}$
hydrogen sulfite	$\text{HSO}_3^-$	thiocyanate	$\text{SCN}^-$	thiosulfate	$\text{S}_2\text{O}_3^{2-}$

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hydrogen	$\text{H}^+, \text{H}^-$	oxygen	$\text{O}^{2-}$	fluorine	$\text{F}^-$
lithium	$\text{Li}^+$	sulfur	$\text{S}^{2-}$	chlorine	$\text{Cl}^-$
potassium	$\text{K}^+$	calcium	$\text{Ca}^{2+}$	scandium	$\text{Sc}^{3+}$
titanium	$\text{Ti}^{2+}, \text{Ti}^{3+}$	vanadium	$\text{V}^{2+}, \text{V}^{3+}$	chromium	$\text{Cr}^{2+}, \text{Cr}^{3+}$
manganese	$\text{Mn}^{2+}, \text{Mn}^{3+}$	iron	$\text{Fe}^{2+}, \text{Fe}^{3+}$	cobalt	$\text{Co}^{2+}, \text{Co}^{3+}$
nickel	$\text{Ni}^{2+}$	copper	$\text{Cu}^+, \text{Cu}^{2+}$	zinc	$\text{Zn}^{2+}$
bromine	$\text{Br}^-$	rubidium	$\text{Rb}^+$	strontium	$\text{Sr}^{2+}$
silver	$\text{Ag}^+$	cadmium	$\text{Cd}^{2+}$	tin	$\text{Sn}^{2+}, \text{Sn}^{4+}$
iodine	$\text{I}^-$	cesium	$\text{Cs}^+$	barium	$\text{Ba}^{2+}$
mercury	$\text{Hg}^{2+}$	lead	$\text{Pb}^{2+}, \text{Pb}^{4+}$		